Primer dilutions

Stock solution

Use #nmoles from the primer specification sheet.

For $100\mu M$: add (#nmoles \times 10) μL of water

For $200\mu M$: add (#nmoles \times 5) μL of water

Working solution

For $4 \mu M$:

$$C_1V_1 = C_2V_2$$

 $100\mu M$ (stock concentration) \times **X** (stock volume needed) = 4 μM (working concentration) \times 200 μL (working volume needed)

$$\boldsymbol{X} = \left(4~\mu M \times 200~\mu L\right)/~100 \mu M$$

$$X = 8 \mu L$$

To prepare working solution:

8 μL (stock solution) + 192 μL (water) = $\frac{200 \mu L}{\mu}$ (working volume needed)